organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Tao Peng,^{a,b} Sheng-Jiao Yan,^{a,b} Chu-Yi Yu^a* and Zhi-Tang Huang^a*

^aBeijing National Laboratory for Molecular Science (BNLMS), Laboratory for Chemical Biology,Institute of Chemistry, Chinese Academy of Sciences, Beijing 100080, People's Republic of China, and ^bGraduate University of The Chinese Academy of Sciences, Beijing 100080, People's Republic of China

Correspondence e-mail: yucy@iccas.ac.cn, huangzt@public.bta.net.cn

Key indicators

Single-crystal X-ray study T = 294 KMean $\sigma(\text{C}-\text{C}) = 0.002 \text{ Å}$ R factor = 0.037 wR factor = 0.111 Data-to-parameter ratio = 12.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

9-(4-Methyoxylbenzoyl)-1,2,3,4,7,8-hexahydropyrido[1,2-*a*]pyrimidin-6-one

In the title compound, $C_{16}H_{18}N_2O_3$, there is an intramolecular $N-H\cdots O$ hydrogen bond. Intermolecular $C-H\cdots O$ bonds stabilize the crystal packing.

Received 13 June 2006 Accepted 20 June 2006

Comment

Heterocyclic ketene aminals (HKAs) (Huang & Wang, 1994), also known as cyclic 1,1-enediamines, are versatile synthons for heterocyclic synthesis. The title compound, (I) (Fig. 1), which possesses a pyrimidine ring, is a member of this family. The crystallographic data for (I) will provide valuable information for assessing its electronic conjugation properties and a possible intramolecular hydrogen bond (Wang *et al.*, 1987), which may be correlated with the reactivity of the secondary amine and the α -C atom (Huang & Wamhoff, 1984).



The C9=O2 double bond [1.2710 (19) Å] is longer than C8=O1 [1.2162 (18) Å]. The nominal C4=C5 double bond [1.402 (2) Å] has almost the same length as the nominal single bond C5-C9 [1.413 (2) Å]. On the other hand, the C4-N2 bond [1.3266 (19) Å] is shorter than a normal C-N single bond. The molecular conformation is stabilized by an intramolecular hydrogen bond $[N2-H2 = 0.86 \text{ Å}, H2\cdots O2 = 1.87 \text{ Å}, N2\cdots O2 = 2.569 (1) \text{ Å}$ and N2-H2 $\cdots O2 = 137^{\circ}$]. The crystal packing of (I) is stabilized by intermolecular C-H \cdots O hydrogen bonds (Fig. 2).



Figure 1

© 2006 International Union of Crystallography All rights reserved View of (I) showing 30% displacement ellipoids (spheres of arbitrary radius for the H atoms). The intramolecular hydrogen bond is indicated by a dashed line.

Experimental

Compound (I) was prepared according to the procedure reported by Zhao *et al.* (1993) and recrystallized from ethyl acetate in 91% yield (m.p. 456 K). Analysis calculated for $C_{16}H_{18}N_2O_3$: C 67.11, H 6.34, N 9.79%; found: C 67.08, H 6.35, N 9.76%.

 $V = 696.6 (3) \text{ Å}^3$

 $D_x = 1.365 \text{ Mg m}^{-3}$

 $0.20 \times 0.18 \times 0.14 \text{ mm}$

3539 measured reflections

2435 independent reflections

1890 reflections with $I > 2\sigma(I)$

Mo $K\alpha$ radiation

 $\mu = 0.10 \text{ mm}^-$

T = 294 (2) K

Prism, yellow

 $R_{\rm int} = 0.028$

 $\theta_{\rm max} = 25.0^\circ$

Z = 2

Crystal data

 $\begin{array}{l} C_{16}H_{18}N_2O_3\\ M_r = 286.32\\ \text{Triclinic, }P\overline{1}\\ a = 7.5696 \ (17) \ \text{\AA}\\ b = 9.725 \ (2) \ \text{\AA}\\ c = 10.446 \ (2) \ \text{\AA}\\ \alpha = 88.964 \ (4)^\circ\\ \beta = 77.971 \ (4)^\circ\\ \gamma = 68.169 \ (3)^\circ \end{array}$

Data collection

Bruker SMART CCD area detector diffractometer φ and ω scans Absorption correction: multi-scan (*SADABS*; Bruker, 2002) $T_{\min} = 0.678, T_{\max} = 1.000$

Refinement

-	
Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0568P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.037$	+ 0.1302P]
$wR(F^2) = 0.111$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.01	$(\Delta/\sigma)_{\rm max} < 0.001$
2435 reflections	$\Delta \rho_{\rm max} = 0.21 \text{ e} \text{ Å}^{-3}$
192 parameters	$\Delta \rho_{\rm min} = -0.14 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	Extinction correction: SHELXL97
	Extinction coefficient: 0.052 (7)

H atoms were positioned geometrically and refined as riding, with C-H = 0.93-0.97 Å and N-H = 0.86 Å; $U_{iso}(H) = 1.2U_{eq}(C,N)$, or $1.5U_{eq}(C)$ for methyl atoms.



Figure 2

The molecular packing of (I), viewed along the *c* axis. H atoms bonded to C atoms have been omitted for clarity. Dashed lines indicate the hydrogen-bonding interactions.

Data collection: *SMART* (Bruker, 2002); cell refinement: *SAINT* (Bruker, 2002); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2002) and *MERCURY* (Bruno *et al.*, 2002); software used to prepare material for publication: *SHELXTL*.

Dr Hai-Bing Song of Nankai University collected the data and is gratefully acknowledged.

References

Bruker (2002). SMART, SAINT, SADABS and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.

- Bruno, I. J., Cole, J. C., Edgington, P. R., Kessler, M., Macrae, C. F., McCabe, P., Pearson, J. & Taylor, R. (2002). Acta Cryst. B58, 389–397.
- Huang, Z. T. & Wamhoff, H. (1984). Chem. Ber. 117, 1856-1867.
- Huang, Z. T. & Wang, M.-X. (1994). Heterocycles, 37, 1233-1262.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Wang, X. J., Zhu, N. J., Guo, F., Liu, Z. R. & Huang, Z. T. (1987). J. Struct. Chem. 6, 62–65.
- Zhao, M. X. & Huang, Z. T. (1993). Synth. Commun. 23, 1039-1046.